## COMBINED SCIENCE

0653/41
Paper 4 Extended Theory
May/June 2017
MARK SCHEME
Maximum Mark: 80

## Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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| Question | Answer | Marks |
| :---: | :--- | :---: |
| 1(a) | lines drawn from Flowering plants to <br> produce haploid pollen ; <br> use auxins to respond to light ; <br> have root hair cells which increase water uptake ; | $\mathbf{3}$ |
| 1(b)(i) | (anthers) <br> hang outside the flower so pollen is easily picked up by wind ; <br> (stigmas) <br> feathery / large surface area to collect pollen ; | $\mathbf{2}$ |
| 1(b)(ii) | to increase the chances of pollination between plants ; |  |
| 1(c)(i) | the idea that fossil fuels contain sulfur/sulfur compounds ; <br> sulfur dioxide produced ; <br> dissolves in water in the air ; | Max 2 |
| 1(c)(ii) | reduces activity / denatures enzymes present in the plants ; | $\mathbf{1}$ |


| Question | Answer | Marks |
| :---: | :---: | :---: |
| 2(a)(i) | covalent ; | 1 |
| 2(a)(ii) | four shared pairs between C and four H atoms ; all symbols correctly shown ; | 2 |
| 2(b)(i) | carbon dioxide ; water ; | 2 |
| 2(b)(ii) | releases heat / thermal energy when it reacts / burns / is used ; | 1 |
| 2(c)(i) | $\underline{\text { natural gas ; }}$ | 1 |
| 2(c)(ii) | coal and petroleum ; | 1 |

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| Question | Answer | Marks |
| :---: | :---: | :---: |
| 3(a)(i) | $\begin{aligned} & \mathrm{D} \\ & \mathbf{C} \end{aligned}$ | 1 |
| 3(a)(ii) | (Force C is 1200 N ) no mark no vertical motion/forces (A and C) are balanced ; | 1 |
| 3(b) | line starts along the speed $=2 \mathrm{~m} / \mathrm{s}$ horizontal, levelling off at speed $=4.5 \mathrm{~m} / \mathrm{s}$ and 10 mins ; any curved line between these points, then level after $(10,4.5)$; | 2 |
| 3(c)(i) | $\begin{aligned} & \mathrm{KE}=1 / 2 \mathrm{mv}^{2} / 1 / 2 \times 120 \times 3 \times 3 ; \\ & =540(\mathrm{~J}) ; \end{aligned}$ | 2 |
| 3(c)(ii) | ( $90 \mathrm{~kJ}=$ ) 90000 J (= work done = energy transferred) ; <br> distance moved $=3(\mathrm{~m} / \mathrm{s}) \times 50(\mathrm{~s})=150 \mathrm{~m}$; <br> force $=$ work done $\div$ distance $/ 90000 \div 150 /=600(\mathrm{~N})$; | 3 |

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| Question | Answer | Marks |
| :---: | :--- | :---: |
| 4(a)(i) | contain chlorophyll ; <br> trap light (energy); <br> converts (light) into chemical energy ; <br> the idea that chemical energy is contained in glucose / starch / carbohydrate ; <br> 4(a)(ii) <br> flagellum ; <br> the idea that the flagellum is for movement ; | $\mathbf{2}$ |
| 4(b)(i) | food chain containing the following organisms <br> phytoplankton $\rightarrow$ zooplankton $\rightarrow$ mussel $\rightarrow$ crab $\rightarrow$ seagull ; <br> four arrows in correct direction in the chain ; | $\mathbf{2}$ |
| 4(b)(ii) | fewer steps / stages / organisms in chain containing mussels / ora ; <br> use of the term trophic level ; <br> energy is lost at each stage ; <br> by heat / movement / avp ; | $\mathbf{3}$ |

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| Question | Answer | Marks |
| :---: | :--- | :---: |
| 5(a)(i) | increases ; <br> neutralisation / salt-making; | $\mathbf{2}$ |
| 5(a)(ii) | CuSO $_{4} ;$ <br> $\mathrm{CO}_{2}$ and $\mathrm{H}_{2} \mathrm{O} ;$ | $\mathbf{2}$ |
| 5(b) | filter (to remove excess solid / copper carbonate) ; <br> heat the solution/filtrate / mixture ; <br> reference to evaporation ; <br> cool /leave (to allow crystals to form) ; | $\mathbf{2}$ |
| 5(c) | the idea that the gradient decreases ; <br> the eidea that the rate decreases ; <br> the idea that the rate becomes zero ; | $\mathbf{2}$ |
| 5(d)(i) | less steep initial line ; <br> levels off at a lower volume; | $\mathbf{2}$ |
| 5(d)(ii) | (decreases rate of reaction) because particles collide less frequently /owtte ; | $\mathbf{1}$ |

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| Question | Answer | Marks |
| :---: | :--- | :---: |
| $7(\mathrm{a})$ | energy $=37 \times 11+17+13 \times 17(=645) ;$ <br> $\times 2=1290(\mathrm{~kJ}) ;$ | $\mathbf{2}$ |
| 7 (b) | eggs (no mark) <br> contains the most fat ; | $\mathbf{1}$ |
| 7 (c)(i) | $6 \mathrm{CO}_{2}$ and $6 \mathrm{H}_{2} \mathrm{O}$; | $\mathbf{1}$ |
| 7 (c)(ii) | in red (blood) cells ; <br> by haemoglobin ; <br> red cells carried in plasma ; | Max $\mathbf{2}$ |
| 7 (d) | chemical digestion:- <br> mouth, and stomach and small intestine / duodenum / ileum ; <br> absorption:- <br> small intestine / duodenum / ileum ; | $\mathbf{2}$ |


| Question | Answer | Marks |
| :---: | :---: | :---: |
| 8(a) | 2,8,3; | 1 |
| 8(b)(i) | oxygen (gas) ; | 1 |
| 8(b)(ii) | aluminium ions gain electrons; gain 3 electrons / ions are discharged / become aluminium atoms; | 2 |
| 8(c)(i) | carbon / C / carbon monoxide / CO ; | 1 |
| 8(c)(ii) | aluminium / Al is more reactive than carbon / C ; | 1 |
| 8(d)(i) | $A l$ is less reactive than $M g$; | 1 |
| 8(d)(ii) | Al is more reactive than Cu ; | 1 |


| Question | Answer | Marks |
| :---: | :--- | :---: |
| $9(a)$ | any two from <br> one or two metals or alloys (other than copper) <br> graphite / carbon | $\mathbf{1}$ |
| $9(\mathrm{~b})$ | $=2 / 0.5=4 ;$ <br> ohms $/ \Omega ;$ | $\mathbf{2}$ |
| $9(\mathrm{c})($ (i) | (2 A) <br> sum of currents in parallel branches = current from source ; | $\mathbf{1}$ |
| $9(\mathrm{c})(\mathrm{ii})$ | $\mathbf{P}$ and $\mathbf{Q}$ have different resistances / thicknesses ; <br> $\mathbf{P}$ less resistance than $\mathbf{Q} / \mathbf{P}$ is thicker than $\mathbf{Q} ;$ | $\mathbf{2}$ |

